Chemistry of PETROCHEMICAL PROCESSES

Second Edition

Provides Quick and Easy Access to Hundreds of Reactions, Processes and Products

Sami Matar Lewis F. Hatch This book is dedicated to the memory of Professor Lewis Hatch (1912–1991), a scholar, an educator, and a sincere friend.

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Contents

Preface to Second Edition xi Preface to First Edition
CHAPTER ONE Primary Raw Materials for Petrochemicals
Natural Gas 1 Natural Gas Treatment Processes 3, Natural Gas Liquids 8, Properties of Natural Gas 10
Crude Oils 11 Composition of Crude Oils 12, Properties of Crude Oils 19, Crude Oil Classification 21
Coal, Oil Shale, Tar Sand, and Gas Hydrates 22
References 26
CHAPTER TWO Hydrocarbon Intermediates
Paraffinic Hydrocarbons 29 Methane 30, Ethane 30, Propane 31, Butanes 31
Olefinic Hydrocarbons 32 Ethylene 32, Propylene 33, Butylenes 34
Dienes 36 Butadiene 37, Isoprene 37
Aromatic Hydrocarbons 37 Extraction of Aromatics 38
Liquid Petroleum Fractions and Residues 42 Naphtha 43, Kerosine 45, Gas Oil 46, Residual Fuel Oil 47
References 47

CHAPTER THREE Crude Oil Processing and Production of Hydrocarbon Intermediates
Introduction 49
Physical Separation Processes 49 Atmospheric Distillation 50, Vacuum Distillation 51, Absorption Process 52, Adsorption Process 52, Solvent Extraction 53
Conversion Processes 54 Thermal Conversion Processes 55, Catalytic Conversion Processes 60
Production of Olefins 91 Steam Cracking of Hydrocarbons 91, Production of Diolefins 101
References 107
CHAPTER FOUR Nonhydrocarbon Intermediates
Hydrogen 111
Sulfur 114 Uses of Sulfur 116, The Claus Process 116, Sulfuric Acid 117
Carbon Black 118 The Channel Process 119, The Furnace Black Process 119, The Thermal Process 119, Properties and Uses of Carbon Black 120
Synthesis Gas 121 Uses of Synthesis Gas 123
Naphthenic Acids 130 Uses of Naphthenic Acid and Its Salts 130
Cresylic Acid 131 Uses of Cresylic Acid 133
References 133
CHAPTER FIVE Chemicals Based on Methane
Chemicals Based on Direct Reactions of Methane 136 Carbon Disulfide 136, Hydrogen Cyanide 137, Chloromethanes 138

Chemicals Based on Synthesis Gas 143 Ammonia 144, Methyl Alcohol 149, Oxo Aldehydes and Alcohols 163, Ethylene Glycol 166
References 167
CHAPTER SIX Ethane and Higher Paraffins-Based Chemicals 169 Introduction 169
Ethane Chemicals 169
Propane Chemicals 171 Oxidation of Propane 171, Chlorination of Propane, 172, Dehydrogenation of Propane 172, Nitration of Propane 173
n-Butane Chemicals 174 Oxidation of n-Butane 175, Aromatics Production 177, Isomerization of n-Butane 180
Isobutane Chemicals 180
Naphtha-Based Chemicals 181
Chemicals from High Molecular Weight n-Paraffins 182 Oxidation of Paraffins 183, Chlorination of n-Paraffins 184, Sulfonation of n-Paraffins 185, Fermentation Using n-Paraffins 185
References 186
CHAPTER SEVEN Chemicals Based on Ethylene
Oxidation of Ethylene 189 Derivatives of Ethylene Oxide 192, Acetaldehyde 198, Oxidative Carbonylation of Ethylene 201
Chlorination of Ethylene 201 Vinyl Chloride 202, Perchloro- and Trichloroethylene 203
Hydration of Ethylene 204
Oligomerization of Ethylene 205 Alpha Olefins Production 206, Linear Alcohols 207, Butene-l 209
Alkylation Using Ethylene 210
References 211

CHAPTER EIGHT Chemicals Based on Propylene
Oxidation of Propylene 214 Acrolein 215, Mechanism of Propene Oxidation 215, Acrylic Acid 217, Ammoxidation of Propylene 218, Propylene Oxide 221
Oxyacylation of Propylene 226
Chlorination of Propylene 226
Hydration of Propylene 227 Properties and Uses of Isopropanol 228
Addition of Organic Acids to Propene 232
Hydroformylation of Propylene: The Oxo Reaction 232
Disproportionation of Propylene (Metathesis) 234
Alkylation Using Propylene 235
References 236
CHAPTER NINE C ₄ Olefins and Diolefins-Based Chemicals
Chemicals from n-Butenes 238 Oxidation of Butenes 239, Oligomerization of Butenes 248
Chemicals from Isobutylene 249 Oxidation of Isobutylene 250, Epoxidation of Isobutylene 251, Addition of Alcohols to Isobutylene 252, Hydration of Isobutylene 253, Carbonylation of Isobutylene 255, Dimerization of Isobutylene 255
Chemicals from Butadiene 255 Adiponitrile 256, Hexamethylenediamine 257, Adipic Acid 257, Butanediol 258, Chloroprene 258, Cyclic Oligomers of Butadiene 259
References 260
CHAPTER TEN Chemicals Based on Benzene, Toluene, and Xylenes
Reactions and Chemicals of Benzene 262

of Benzene 278, Oxidation of Benzene 280, Hydrogenation of Benzene 281
Reactions and Chemicals of Toluene 284 Dealkylation of Toluene 284, Disproportionation of Toluene 285, Oxidation of Toluene 286, Chlorination of Toluene 291, Nitration of Toluene 292, Carbonylation of Toluene 294
Chemicals from Xylenes 294 Terephthalic Acid 295, Phthalic Anhydride 296, Isophthalic Acid 297
References 299
CHAPTER ELEVEN Polymerization
Monomers, Polymers, and Copolymers 302
Polymerization Reactions 303 Addition Polymerization 304, Condensation Polymerization 312, Ring Opening Polymerization 314
Polymerization Techniques 315
Physical Properties of Polymers 317 Crystallinity 317, Melting Point 317, Viscosity 317, Molecular Weight 318, Classification of Polymers 320
References 321
CHAPTER TWELVE Synthetic Petroleum-Based Polymers
Thermoplastics 324 Polyethylene 324, Polypropylene 329, Polyvinyl Chloride 332, Polystyrene 334, Nylon Resins 336, Thermoplastic Polyesters 336, Polycarbonates 337, Polyether Sulfones 339, Poly(phenylene) Oxide 340, Polyacetals 341
Thermosetting Plastics 342 Polyurethanes 342, Epoxy Resins 344, Unsaturated Polyesters 346, Phenol-Formaldehyde Resins 346, Amino Resins 348

Synthetic Rubber 350
Butadiene Polymers and Copolymers 352, Nitrile Rubber 353, Polyisoprene 354, Polychloroprene 356, Butyl Rubber 356, Ethylene
Propylene Rubber 357, Thermoplastic Elastomers 358
Synthetic Fibers 359
Polyester Fibers 359, Polyamides 362, Acrylic and Modacrylic Fibers 368, Carbon Fibers 369, Polypropylene Fibers 370
References 371
Appendix One: Conversion Factors
Appendix Two: Selected Properties of Hydrogen, Important C_1 – C_{10} Paraffins, Methylcyclopentane, and Cyclohexane 376
Index
About the Authors392

Preface to Second Edition

When the first edition of *Chemistry of Petrochemical Processes* was written, the intention was to introduce to the users a simplified approach to a diversified subject dealing with the chemistry and technology of various petroleum and petrochemical process. It reviewed the mechanisms of many reactions as well as the operational parameters (temperature, pressure, residence times, etc.) that directly effect products' yields and composition. To enable the readers to follow the flow of the reactants and products, the processes were illustrated with simplified flow diagrams.

Although the basic concept and the arrangement of the chapters is this second edition are the same as the first, this new edition includes many minor additions and updates related to the advances in processing and catalysis.

The petrochemical industry is a huge field that encompasses many commercial chemicals and polymers. As an example of the magnitude of the petrochemical market, the current global production of polyolefins alone is more than 80 billion tons per year and is expected to grow at a rate of 4–5% per year. Such growth necessitates much work be invested to improve processing technique and catalyst design and ensure good product qualities. This is primarily achieved by the search for new catalysts that are active and selective. The following are some of the important additions to the text:

Because ethylene and propylene are the major building blocks for petrochemicals, alternative ways for their production have always been sought. The main route for producing ethylene and propylene is steam cracking, which is an energy extensive process. Fluid catalytic cracking (FCC) is also used to supplement the demand for these light olefins. A new process that produces a higher percentage of light olefins than FCC is deep catalytic cracking (DCC), and it is described in Chapter 3.

- The search for alternative ways to produce monomers and chemicals from sources other than oil, such as coal, has revived working using Fisher Tropseh technology, which produces in addition to fuels, light olefins, sulfur, phenols, etc. These could be used as feedstocks for petrochemicals as indicated in Chapter 4.
- Catalysts for many petroleum and petrochemical processes represent a substantial fraction of capital and operation costs. Heterogeneous catalysts are more commonly used due to the ease of separating the products. Homogeneous catalysts, on the other hand, are normally more selective and operate under milder conditions than heterogeneous types, but lack the simplicity and ease of product separation. This problem has successfully been solved for the oxo reaction by using rhodium modified with triphenylphosphine ligands that are water soluble. Thus, lyophilic products could be easily separated from the catalyst in the aqueous phase. A water soluble cobalt cluster can effectively hydroformylate higher olefins in a two-phase system using polyethylene glycol as the polar medium. This approach is described in Chapter 5.
- In the polymer filed, new-generation metallocenes, which are currently used in many polyethylene and polypropylene processes, can polymerize proplylene in two different modes: alternating blocks of rigid isotactic and flexible atactic. These new developments and other changes and approaches related to polymerization are noted in Chapters 11 and 12.

I hope the new additions that I felt necessary for updating this book are satisfactory to the readers.

Sami Matar. Ph.D.